



*[U- $^{13}\text{C}$ ] succinate*

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**Field strength:** 14.1 T

### Why is this your favorite spectrum?

Succinate is a very innocent-looking simple molecule. Despite this, uniformly labelled succinate results in quite complex  $^{13}\text{C}$  NMR spectra. Strong coupling between the two central carbon nuclei leads to  $^1\text{H}$  decoupled spectra (shown in red) of the  $\text{COOH}$  and the  $\text{CH}_2$  carbon nuclei which are, apart from the different intensity due to the  $^1\text{H}$  decoupling, very similar. This similarity is not observable in the fully coupled spectra (shown in blue). I think this is a fantastic example showing the wealth of information contained in NMR spectra.