



$[U-^{13}\text{C}]$ succinate

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Field strength: 14.1 T

Why is this your favorite spectrum?

Succinate is a very innocent-looking simple molecule. Despite this, uniformly labelled succinate results in quite complex ^{13}C NMR spectra. Strong coupling between the two central carbon nuclei leads to ^1H decoupled spectra (shown in red) of the COOH and the CH₂ carbon nuclei which are, apart from the different intensity due to the ^1H decoupling, very similar. This similarity is not observable in the fully coupled spectra (shown in blue). I think this is a fantastic example showing the wealth of information contained in NMR spectra.